CMGT 235 – Electrical and Mechanical Systems

Unit 1 - Mechanical Systems

Fall 2022

Discussion No. 3 Properties of Air-Water Mixtures

Definitions and Concepts

- HVAC Heating, Ventilation, and Air conditioning
- Psychrometrics The behavior of mixtures of air and water vapor under varying conditions of heat
- Enthalpy = Total heat in the air = Sensible plus Latent heat
 - Sensible Heat Changes in temperature that do not alter the moisture content of air
 - Latent Heat Related to level of moisture in the air
- BTU (British Thermal Unit) The amount of heat that must be added to or subtracted from a pound of water at 60°F to affect a temperature change of 1°F
 - BTUH or BTU/H BTU's per hour
 - MBTUH 1000 BTUH
- Ton
 - 1 Ton equals the amount of heat needed to melt 1 ton of ice in one day
 - 12,000 BTUH's
- Drybulb Temperature The temperature reading given by a dry thermometer that gives a direct indication as to the sensible heat content of air
- Wetbulb Temperature The temperature reading from a wetted bulb that gives a direct indication as to the total heat content of air
- Dew Point Temperature Temperature at which air will begin to release moisture.
- Relative Humidity (RH)
 - The actual amount of moisture in the air expressed as a percentage of the amount of moisture the air is capable of holding.

More technically:

- The amount of water vapor in the air divided by the amount of water vapor the air can hold (at the same temperature and pressure.)
- The ratio of the air's vapor pressure to its saturation vapor pressure.
- Example: An air sample that is at 50% RH is holding half the moisture it is capable of holding at the same temperature (at dew point or saturated.)
- RH is inversely relational to temperature for the same moisture level (grains of moisture per pound of dry air) warm air can hold more moisture
- RH is what we sense
- High RH: Sticking, mold
- Low RH
 - Affects electronics, promotes static
 - o Low RH air is seeking saturation, absorbing moisture wherever it can
- Specific Humidity or Humidity Ratio
 - The weight of the water vapor in each pound of dry air
 - Typically grains of moisture/pound of dry air
 - Grain = 1/7000 pound
- Density Unit weight of dry air at a given temperature and moisture content, lb/ft³
- Specific Volume Space occupied by dry air at a given temperature and moisture content (the reciprocal of density), ft³/lb

The Psychrometric Chart [SEE PAGE 4 FOR PSYCHROMETRIC CHART]

- Constant Drybulb Temperature: Vertical Lines
- Constant Wetbulb temperature: Upward left sloping lines
- Relative humidity: Curving lines (100% line is the saturation curve or correlates with Dew Point)
- Left hand scale: Constant Dew Point (°F):
 Right hand scale: Humidity Ratio (grains of moisture/lb of dry air) and Vapor Pressure (in of Hg)
- Enthalpy or total heat, BTU/lb of dry air: Staggered scale left of saturation curve and left sloping lines
- Constant specific volume, ft³/lb of dry air: Sloping lines
- Saturation Curve: 100% RH Curve (or the point at which an air mixture can hold no additional moisture at a given temperature); temperature on the curve is the Dew Point

Latent vs. Sensible Changes

- Latent Changes move in the direction of the y-axis
- Sensible changes move in the direction of the x-axis

Relationship of Drybulb (DB), Relative Humidity (RH), Dewpoint (DP), and Wetbulb (WB)

Impact to Relative Humidity from Sensible Changes

Temperature and Relative Humidity are inversely related:

- the higher the temperature, the lower the RH
- the lower the temperature, the higher the RH

Example 1. See Psychrometric Chart, next page

If the air temperature of 75°F at 40% RH is warmed to 90°F without adding moisture, what is the new RH? RH = 25%

What is the RH is the air is cooled to $55^{\circ}F$? RH = 80%

Dehumidification

What happens when the RH is 100%? It Rains! Saturation — As air cools, it eventually reaches its dewpoint and moisture begins to appear.

If the air is continued to be cooled, at what temperature will the saturation point be reached? $DB = 49^{\circ}F$

What happens if the air is continued to be cooled below 49°F? Cooling below 49°F removes water i.e. The air can be de-humidified by over cooling it.

Humidification

If the air temperature of 75°F at 40% RH what is the new RH if 12 gr/lb of moisture is added to the air? RH = 50%

What is the new DP? DP = 55°F

Enthalpy Changes

- Enthalpy = Total heat in the air = Sensible plus Latent heat
 - Sensible Heat Changes in temperature that do not alter the moisture content of air
 - Latent Heat Related to level of moisture in the air
- Enthalpy is measured in BTU/lb dry air

What is the change in enthalpy when air at 75°F at 40% RH changes to 80°F at 60% RH?

@75°F H_i = 26.2 BTU/lb DA

 $@80^{\circ}F$ H_f = 33.8 BTU/lb DA

 Δ Enthalpy = Δ H = 33.8 - 26.2 = 7.6 BTU/lb DA

Example 2.

A space at sea level is 70°F and RH = 50%. Find the other properties of the air in that space. Solution. Use Psychrometric Chart

| DB = | 70°F |
|-----------------|-------------------------------------|
| RH = | 50% |
| WB = | 58.5°F |
| Humidity Ratio= | 55 gr/lb of dry air |
| VP = | 0.37 in Hg |
| Enthalpy = | 25.5 BTU/lb of dry air |
| Vs = | 13.5 ft ³ /lb of dry air |
| Dew Point = | 50.5°F |

Example 3.

A saturated mixture contains 100 lb of dry air. How much heat is required (Btu) to raise the dry bulb temperature from 30°F to 70°F?

Solution. Use Psychrometric Chart

@30°F H_i = 10.9 BTU/lb DA

@70°F H_f = 20.8 BTU/lb DA

 $\Delta H = 20.8 - 10.9 = 9.9 \text{ BTU/lb DA}$

 $Q = m \Delta H = 100 \text{ lb x } 9.9 \text{ BTU/lb } DA = 990 \text{ BTU}$